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Ch 6 Study Guide answers

CHAPTER 6 REVIEW Chemical Bonding SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. a The notation for sodium chloride, NaCl, stands for one (a) formula unit. (c) crystal. (b) molecule. (d) atom. 2. d In a crystal of an ionic compound, each cation is surrounded by a number of (a) molecules. (c) dipoles. (b) ...

6 Chemical Bonding - Effingham County Schools / Overview

Modern Chemistry Chapter 6 Chemical Bonding . Chemical Bond •A link between atoms that results from the mutual attraction of their nuclei for electrons –Electrostatic attraction between proton and electron –Classified by the way the valence e- are distributed around nuclei of combined atoms .

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CHAPTER 6 REVIEW Chemical Bonding

Modern Chemistry • CHAPTER 6 HOMEWORK 6-3 (pp. 167-169) VOCABULARY Write true or false for each statement. 1. The bond length is the distance between two bonded atoms at their maximum potential energy. ____ 2. The energy required to form a chemical bond between two neutral atoms is called bond energy. ____ 3.

HOMEWORK 6-1

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Modern Chemistry 47 Chemical Bonding CHAPTER 6 REVIEW Chemical Bonding SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. ____ In metals, the valence electrons are considered to be (a) attached to particular positive ions.

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Modern Chemistry Chapter 6 Review Answers Section 3

Modern Chemistry 33 .Name Section Quiz, continued Class Date 6. The Lewis structure of which of the following compounds does not follow the octet rule? a. NH3 b BF3 c. d. ... each statement or best answers each question. 1. Chemical bonding in metals is a. the same as ionic bonding. b. the same as covalent bonding.

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CHAPTER 6 Chemical Bonding

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