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Corrosion is a galvanic process that can be prevented using cathodic protection. The deterioration of metals through

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oxidation is a galvanic process called corrosion. Protective coatings consist of a second metal that is more difficult to oxidize than the metal being protected.

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Denny A. Jones is the
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To be an effective corrosion site, the crevice must be wide enough to allow the corroding electrolyte in and then provide 'stagnant' conditions. Crevice corrosion can therefore be a concern where gaps are a few micrometers wide, but there are no absolute or critical dimensions for crevices, below which corrosion is

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certain.

Article: Principles and prevention of crevice corrosion

Thus it is critical to use corrosion prevention methods and use Zavenir products to halt, stop and prevent corrosion in metals.

The prevention can be done through cathodic protection where through using different currents the undesirable anodic

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sites on a metals

surface is converted to
cathodic destination.

The next way is coating
the metal.

Different Corrosion Prevention Method

Prevention of Corrosion

Preventing corrosion is
of utmost importance
in order to avoid huge
losses. Majority of the
structures we use are
made out of metals.

This includes bridges,
automobiles,

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machinery, household goods like window grill, doors, railway lines, etc.

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electrochemical
process in which
metals and alloys
undergo
transformation into
predominantly oxides,
hydroxides, and
aqueous salts. In the
corrosion process, two
reactions take place.

Corr Science » Principles of Corrosion

Cathodic protection is
a proven corrosion
control method for

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protection of

underground and

undersea metallic

structures, such as oil

and gas pipelines,

cables, utility lines, and

structural foundations.

The chapter illustrates

the simple principle of

cathodic protection.

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presents a functional

approach to the

various forms of

corrosion, such as

uniform corrosion,

pitting corrosion,

crevice corrosion,

galvanic corrosion,

stress corrosion,

hydrogen-induced

damage, sulphide

stress cracking, erosion-

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corrosion, and
corrosion fatigue in
various industrial
environments.

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